

# Evolution of HCl Concentrations in the Lower Stratosphere from 1991 to 1996 Following the Eruption of Mt. Pinatubo

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**In situ measurements of hydrochloric acid (HCl) concentrations in the lower stratosphere from 1991 to 1996 reveal that its abundance relative to that of total inorganic chlorine ( $\text{Cl}_\text{t}$ ) has evolved upwards from  $\text{HCl}/\text{Cl}_\text{t} = 40\%$  in late 1991 to 70% in 1996. The observed fraction is anticorrelated with aerosol surface area concentration, which has been diminishing since June 1991 following the volcanic eruption of Mt. Pinatubo. For several years, calculations from photochemical models including heterogeneous chemistry predicted little change in  $\text{HCl}/\text{Cl}_\text{t}$ , and were unable to reproduce these observations. However, calculations incorporating new laboratory results of faster heterogeneous chemistry show that air parcels with high aerosol loading exposed to temperatures of 200-210 K experience enough chlorine activation to drive the  $\text{HCl}/\text{Cl}_\text{t}$  fraction below 60%, as observed. This repartitioning of  $\text{Cl}_\text{t}$  increases ozone loss at all latitudes.**

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## Introduction

The massive enhancement in aerosol loading of the lower stratosphere following the June 1991 eruption of Mt. Pinatubo [McCormick *et al.*, 1995; Jonsson *et al.*, 1996] provided an opportunity to test and develop our understanding of the role of heterogeneous chemistry on sulfate aerosol particles [Jones *et al.*, 1994]. As the dense aerosol volcanic cloud spread into a relatively clean atmosphere, aircraft, balloon, ground-based, and space measurements reported large reductions in concentrations of stratospheric  $\text{NO}_\text{x}$  ( $\text{NO} + \text{NO}_2$ ) relative to those of total reactive nitrogen ( $\text{NO}_\text{y}$ ) [Fahey *et al.*, 1993; Koike *et al.*, 1994; Webster *et al.*, 1994a], and in response, a doubling of concentrations of chlorine monoxide ( $\text{ClO}$ ) [Avallone *et al.*, 1993] and chlorine dioxide ( $\text{ClO}_2$ ) [Solomon *et al.*, 1993]. Heterogeneous hydrolysis of dinitrogen pentoxide ( $\text{N}_2\text{O}_5$ ) and chlorine nitrate ( $\text{ClONO}_2$ ) to nitric acid ( $\text{HNO}_3$ ) were identified as the reactions responsible for the observed changes in atmospheric composition. Column measurements of HCl before and after the eruption showed no

significant injection of volcanic HCl into the stratosphere [Mankin *et al.*, 1992], and photochemical model calculations incorporating the heterogeneous hydrolysis of  $\text{N}_2\text{O}_5$  predicted little change ( $\leq 15\%$ ) in stratospheric HCl amounts despite the large impact on  $\text{NO}_x$  and  $\text{ClO}_x$  amounts [Salawitch *et al.*, 1994]. Space-based measurements of either chlorine reservoir HCl or  $\text{ClONO}_2$  were limited by the dense volcanic cloud to altitudes above the high aerosol loading where heterogeneous chemistry was occurring. *In situ* measurements of  $\text{ClONO}_2$  have never been made, and *in situ* measurements of HCl below 25 km have only been available (this work) since late 1991, significantly after the volcanic material had spread.

Since 1992, the stratospheric aerosol loading has decayed considerably; the surface area layer available in 1996 for heterogeneous chemistry has diminished by over an order of magnitude to the low background levels existing before the Pinatubo eruption [McCormick *et al.*, 1995]. Mapping the recovery over the last four years of radical and reservoir abundances to pre-eruption levels offers a rigorous test of models incorporating heterogeneous chemistry, since this period was one of slow reduction in a more homogeneously-distributed aerosol concentration, with more frequent sampling opportunities. During this time, *in situ* aircraft measurements of HCl have been made, and the remote-sensing instruments have been better able to penetrate the volcanic aerosol and sample lower in the stratosphere,

In the lower stratosphere where heterogeneous chemistry has a profound effect on the contribution of the  $\text{ClO}_x$ ,  $\text{NO}_x$ , and  $\text{HO}_x$  catalytic cycles to ozone loss (see the review of Jones *et al.*, [1994]), the relative amount of HCl and  $\text{ClONO}_2$  takes a special importance through chlorine nitrate's role as a temporary reservoir for reactive forms of both chlorine and nitrogen oxides. Unlike HCl,  $\text{ClONO}_2$  is readily photolyzed to produce Cl or ClO. The relative abundances of HCl and  $\text{ClONO}_2$  can therefore determine the extent of ozone loss. Although HCl is recognized as the principal reservoir of chlorine in the middle and upper stratosphere, discrepancies exist between models and measurements of HCl both above 24 km where models tend to under-predict HCl abundances [Michelsen *et al.*, 1996], and below 24 km where models over-predict HCl abundances [Webster *et al.*, 1994 b].

Since 1991, JPL's Aircraft Laser Infrared Absorption Spectrometer (ALIAS) instrument [Webster *et al.*, 1994c] (I5) has flown more than 150 times on the NASA's ER-2 aircraft. HCl mixing ratios first measured by this instrument [Webster *et al.*, 1993] from 15-20 km were about a factor of two lower than model predictions [Salawitch *et al.*, 1994]. Results from the 1993 Stratospheric Photochemistry and Aerosols and Dynamics (SPADE) campaign out of Moffett Field, California, illustrated the difficulties in balancing the inorganic chlorine budget using the low HCl measurements. In the absence of *in situ* measurements of  $\text{ClONO}_2$ , simultaneous measurements of ClO,  $\text{NO}_2$ , and  $\text{O}_3$  were used to infer  $\text{ClONO}_2$  abundances [Stimpfle *et al.*, 1994]. With an inferred fraction of  $\text{ClONO}_2 / \text{Cl}_y$  of 14% (ranging from 6 to 28%), the measured ClO amounts implied that if Cl<sub>y</sub>

comprised only HCl and ClONO<sub>2</sub>, if Cl/Cl<sub>y</sub> should be ~86%, compared to the observed value of 40%. The aircraft measurements of 1993 therefore left ~30% of inorganic chlorine apparently unaccounted for, indicating that the photochemistry involving HCl and ClONO<sub>2</sub> was not well understood. An analysis of data recorded in September 1993 during a balloon flight of the Mark IV FTIR spectrometer yielded the proposal that 20-30% of Cl<sub>y</sub> may be sequestered as perchloric acid (HClO<sub>4</sub>) produced from the heterogeneous reaction of ClO and H<sub>2</sub>SO<sub>4</sub> on sulfate aerosols [Jaeglé *et al.*, 1995]. Two other studies report HCl/Cl<sub>y</sub> ratios consistent with both the model predictions of 80% and with the aircraft ALIAS measurements of 1993. These studies include: balloon measurements of ClO at 20 km [Avallone *et al.*, 1993], and a recent analysis of HALOE HCl and ClAES ClONO<sub>2</sub> measurements from the UARS satellite [Dessler *et al.*, 1995] observed between August 1992 and March 1993, both of which reported HCl/Cl<sub>y</sub> = 60 ± 20 %. In an intercomparison of 1994 aircraft data with overflights of ATMOS in Nov. 1994, and with an earlier flight of Mark IV in May 1994, the ALIAS data lie midway between the two FTIR data sets for N<sub>2</sub>O values ≤ 210 ppbv, where HCl mixing ratios are ≥ 1 ppbv. ALIAS measurements are about 10% lower than that of ATMOS [Chang *et al.*, 1996], and about 10% higher than those of the Mark IV [G. Toon, private communication], for similar ozone abundances.

This paper brings together six years of HCl measurements to show a remarkable consistency in the aircraft data base and its evolution with time from 1991-96. It places earlier HCl measurements in context with those made more recently (1992-96) during the years of gradual decay of the Mt. Pinatubo aerosol loading of the stratosphere. Particularly intriguing about the time history of the aircraft measurements is the observation of an HCl fraction of total inorganic chlorine (Cl<sub>y</sub>), which has increased with time as the Mt. Pinatubo aerosol loading has diminished; values of HCl/Cl<sub>y</sub> range from 40% in 1991/1992 to 70% in 1996, now evolved upward into good agreement with earlier model predictions. We present here model results incorporating new laboratory results of faster heterogeneous chemistry, which show that air masses with high aerosol loading exposed to temperatures typical of the midlatitude lower stratosphere (200-210 K) experience enough chlorine activation to drive the HCl/Cl<sub>y</sub> fraction below 60%, as observed. Significant repartitioning of chlorine reservoir species from low-temperature sulfate aerosol chemistry and enhancement in ozone depletion is therefore predicted at all latitudes from even moderate future volcanic activity.

## The aircraft instruments

The Aircraft Laser Infrared Absorption Spectrometer (ALIAS) instrument is a 4-channel scanning tunable diode laser spectrometer that uses high-resolution laser absorption in the 3-8 μm wavelength region [Webster *et al.*, 1994c] to directly and simultaneously measure any four of the gases HCl, NO<sub>2</sub>, CH<sub>4</sub>, CO, and N<sub>2</sub>O. Both direct absorption and second-harmonic detection techniques are used to measure the fractional absorption from individual or groups of rovibrational lines for each gas, with line-center absorption typically a few percent for the tracers N<sub>2</sub>O, CH<sub>4</sub>, and CO, and less than 0.05% for HCl and NO<sub>2</sub>. The instrument samples the atmosphere using a fast flow system with a flow rate of 16 liters/sec

at 65 mbar driven by inlet ram pressure and by aerodynamic vacuum at the exhaust. The flow system is in two parts, a heated inlet with wall temperatures of 60-70°C, and a multipass optical (Herriott) cell. All interior surfaces are coated with a halocarbon wax from a chloroform solution to minimize wall loss. Extensive calibration and testing of the inlet and sampling system, including post-flight ion-analysis of the inlet residue, has revealed no systematic instrument or sampling error that could result in HCl loss of greater than 10%. The fidelity of the measurement of HCl in the multi-pass cell is checked by simultaneous recording of strong CH<sub>4</sub> lines close to the HCl absorption line. The precision of the data can vary from flight to flight, but is typically a constant value of about 0.05 ppbv. The accuracy of the measurement is about 5-7%, producing total measurement uncertainties of about 10-15% for HCl.

The aerosol number, surface area, and volume concentrations reported here were determined from three particle size spectrometers flown simultaneously as part of the ER-2 aircraft payload, with overall uncertainties of 35% estimated from intercomparison. The condensation nucleus counter (CNC) measured particle concentrations larger than 0.008 µm in diameter [Wilson *et al.*, 1983], with an accuracy of a few percent. The focused cavity aerosol spectrometer (FCAS) instrument measured size distributions from 0.06 to 2 µm diameter range [Wilson *et al.*, 1992], producing a particle size spectrum every 10 seconds. The forward scattering spectrometer probe (FSSP-300) detected particles between (1.4 and 20 µm diameter [Baumgardner *et al.*, 1992], and flew on the ER-2 during 1991/2 and 1993. We used measurements from FSSP for AASE-II, from FCAS for ASHOE/MAESA and STRAT, and from all three spectrometers for SPADE.

For most of the flight data presented here, the tracer N<sub>2</sub>O was measured directly by the ALIAS instrument. However, for a few flights in which ALIAS data were unavailable or of poor quality, N<sub>2</sub>O measurements from the Airborne Tunable Laser Absorption Spectrometer (ATLAS) instrument [Podolske and Loewenstein, 1993] were used. Cl(O) was measured using resonance fluorescence detection of Cl atoms at 118.9 nm, generated by chemical conversion of ambient ClO to Cl by addition of NO [Stimpfle *et al.*, 1994]. O<sub>3</sub> was measured using UV absorption [Proffitt *et al.*, 1983].

## **The aircraft measurements of HCl**

Figure 1 shows plots of HCl vs. N<sub>2</sub>O recorded by the ALIAS instrument as part of the payload of the ER-2 aircraft during four major missions: the 1991/2 Airborne Arctic Stratospheric Expedition (AASE-II), the 1993 SPADE experiment, the 1994 Airborne Southern Hemisphere Ozone Expedition/Measurements of the Atmospheric Effects of Stratospheric Aircraft (ASHOE/MAESA), and the 1996 Stratospheric Tracers of Atmospheric Transport (STRAT) mission. These data are principally from latitudes between 20° and 70° N and have been restricted to pressures less than 67 mbar and ClO amounts less than 150 pptv to remove data from air masses that have undergone severe heterogeneous low-temperature processing. Also plotted are the calculated values of Cl<sub>y</sub>,

expected from the observed  $N_2O$  values and a polynomial least-squares fit to the data.

The 1991 /2 AASE-II data collected from a series of flights from 4 Oct. 1991 through 22 March 1992 fall mainly within a single correlation group except for the striking anomalies from the two earliest flights of 4 Oct. (blue triangles) and 6 Oct. (black diamonds) 1991. These latter two flights covered the latitude range 37-85 °N and were sufficiently close in time to the June 1991 eruption of Mt. Pinatubo to sample regions of rapidly-changing aerosol loading [Jonsson *et al.*, 1992]. The data from these two early flights appear divided into two populations of low and high HCl values, whose variation with aerosol surface area is described later. The 1993 SPADe HCl values are generally more tightly correlated with measured  $N_2O$ , except the flights of 11 May (orange triangles) and 1 May (green circles) 1993, which are not included in the least-squares correlation fit. From tracer correlations, these flights have been identified as sampling air more typical of midlatitudes than of polar origin [Michelsen *et al.*, 1997a] (see next section).

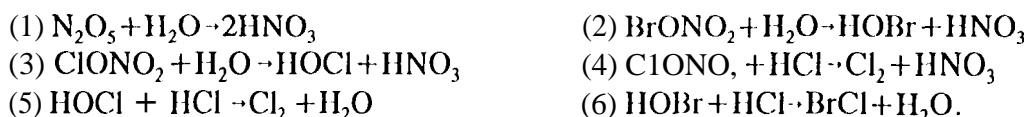
The least-squares fits of Fig. 1 from each mission are plotted in Fig. 2A, where an evolution with time of the HCl/ $Cl_y$  ratio is apparent. The data from 1991/2 (AASE-II), 1993 (SPADe), 1994 (ASHOE/MAESA), and 1996 (STRAT) show HCl/ $Cl_y$  fractions of typically 40%, 50%, 55-60%, and 75%, a trend that is maintained with latitude within the range 40-60 °N, as shown in Fig. 211, where data from all missions are included.

### Chlorine activation on sulfate aerosols

The chemistry responsible for the formation of the ozone hole is initiated by the heterogeneous conversion of reservoir species such as HCl and  $ClONO_2$  to reactive radicals (such as  $Cl$  and  $ClO$ ), a process known to occur on the surfaces of polar stratospheric cloud (PSC) particles and supercooled liquid sulfate aerosols, as described in the review by Fahey *et al.* [1995]. This heterogeneous loss of HCl is associated with low temperatures, i.e., less than 195 K, and is not expected to be significant at higher temperatures.

In a recent study of ATMOS, SAGE 11, and UARS data from the northern hemisphere polar vortex, however, Michelsen *et al.* [1997b] concluded that reactions mediated by liquid sulfate aerosols (60% wt. ) could have a greater effect than previously believed on the midlatitude partitioning of  $Cl_y$  and  $NO_y$  species, and on global ozone loss throughout the lower stratosphere. These reactions were found to repartition  $Cl_y$  and  $NO_y$  at temperatures warmer than 197 K, which are more typical of those existing outside the polar vortex. In a second study, these authors studied correlations of  $CH_4$ ,  $N_2O$ ,  $NO_y$ ,  $H_2O$ , and aerosol abundances from ATMOS, SAGE, and ER-2 data sets to discriminate between vortex, midlatitude, and tropical air masses. From this framework, it was identified [Michelsen *et al.*, 1997a] that much of the air sampled during SPADe had the tracer signatures of polar air, characterized by low  $CH_4$ , low  $NO_y$ , low  $O_3$ , high aerosol surface area, low HCl, and low  $H_2O$  amounts. Aircraft data from other years were not considered.

We have tailored the heterogeneous model results of *Michelsen et al.* [1997a,b] to estimate the sensitivity of the ratio  $\text{HCl}/\text{Cl}_y$  to aerosol surface area concentration and temperature for comparison with the aircraft data, to assess whether heterogeneous sulfate chemistry can quantitatively reproduce the observations of a five-year data base. In Fig. 4A the flight-average  $\text{HCl}/\text{Cl}_y$  ratios vs. aerosol surface concentration for all flights studied are compared with those calculated by a model assuming photochemical steady state and incorporating the full set of heterogeneous reactions:



At temperatures below about 206 K, these reactions proceed rapidly enough on liquid sulfuric acid (60 wt%) aerosol particles to provide a significant channel for Cl production at the expense of HCl, leading to enhanced  $\text{ClONO}_2$  production from the gas-phase reaction of  $\text{NO}_2$  and ClO. In this way the partitioning of  $\text{Cl}_y$  between HCl and  $\text{ClONO}_2$  is shifted from HCl dominance to that of  $\text{ClONO}_2$ . The importance of including heterogeneous bromine chemistry has been emphasized by the recent studies of *Lary et al.* [1996] and *Tie and Brasseur* [1996], whose studies suggested that significant reduction in  $\text{HCl}/\text{Cl}_y$  could be achieved from low temperature heterogeneous bromine chemistry alone. Within the 16-19 km region from which the ALIAS data predominantly originates, our model calculations show that reactions (4) and (5) account for 40-50% each of the repartitioning, with reactions (1), (2), (3), and (6) making little ( $\leq 15\%$ ) contribution.

Since  $\text{HCl}/\text{Cl}_y$  has a significant dependence on  $\text{O}_3$  abundance, we first restrict data to points lying in the range  $1500 \leq \text{O}_3 \leq 2500$  ppbv (with the additional restrictions of  $\text{N}_2\text{O} \leq 260$  ppbv, pressure  $\leq 67$  mbar, ClOS 150 pptv, and  $\text{sza} \leq 82^\circ$ ). A plot of  $\text{HCl}/\text{Cl}_y$  against aerosol surface area concentration shows that with few exceptions (4 and 6 Oct. 1991; 1 and 11 May 1993; included in Fig. 4 and discussed later), flights from 1991-6 appear to demonstrate a smooth decay in  $\text{HCl}/\text{Cl}_y$  with increasing surface area, as shown in Fig. 3. The lower panel is a fit to the flight-average points using an exponential decay function.

The flight averages of Fig. 3 are included with those of 4 and 6 Oct. 1991, and of 1 and 11 May, 1993, for comparison with model calculations run at several temperatures initialized with values of  $\text{NO}_y$ ,  $\text{CH}_4$ ,  $\text{O}_3$ ,  $\text{H}_2\text{O}$ , and  $\text{N}_2\text{O}$  measured during the flight of 2 Nov. 1994. For high surface area concentrations, the observed decay in  $\text{HCl}/\text{Cl}_y$  is generally well-represented in Fig. 4 by the model calculations for a temperature of about 207 K. There is a tendency, however, for the observed decay to be steeper at lower surface areas, and flatter at larger surface areas. A more thorough analysis based on conditions measured for each flight is again more consistent with the measurements than the results of previous modeling studies, but still shows disagreement with measured values when photochemical steady-state is assumed for the instantaneous temperatures measured during

the flight. For the high surface area flight of 17 February 1992, for example, in which temperatures of 210-225 K were encountered with a mean  $\text{HCl}/\text{Cl}_y$  value close to 0.4, flight-tailored model calculations produce  $\text{HCl}/\text{Cl}_y$  values of 0.2, 0.49, and 0.67 for temperatures of 205, 207, and 221 K, respectively. The model results are sensitive to the duration of low temperatures recently encountered, and back-trajectory calculations of the temperatures experienced by the sampled air masses are needed for all of the flights for better comparison with the model results. During the AASIF-II, ASIIO/VMAESA, and STRAT missions, flight temperatures were rarely lower than 210-215 K, but during the SPADE mission flight temperatures in the range 203-208 K were often recorded; this range therefore represents an upper value for the minimum temperatures experienced over the last few days. The flights of 1 and 11 May 1993 were recorded at the higher temperatures of 215-220 K, and 220 K, respectively, and the flight average  $\text{HCl}/\text{Cl}_y$  ratios are seen in Fig. 4 to be higher than the 1992-96 fit. Tracer correlations indicate that these flights should be in photochemical steady-state for midlatitude conditions, consistent with the model results.

Because the combination of high surface area and moderately cold temperature is necessary to repartition  $\text{HCl}$ , an altitude dependence to  $\text{HCl}/\text{Cl}_y$  is expected. Above ER-2 altitudes, temperatures are generally warmer. The pressure-dependent  $\text{HCl}$  vs.  $\text{N}_2\text{O}$  relationship shown by several data sets [Webster *et al.*, 1994b] may reflect the ineffectiveness of aerosol chemistry at the altitudes above the ER-2 where higher temperatures are found, despite the enhanced aerosol loading.

Also plotted in Fig. 4 are the flight-average data for the early flights of 4 and 6 Oct. 1991, which have particularly low  $\text{HCl}/\text{Cl}_y$  ratios for surface area concentrations only 2-5 times background levels. Although the model calculations suggest that these data may be from air masses that had earlier experienced temperatures of 203 K or less, temperatures this low are not generated from back trajectory calculations five days before the flights. Looking back at Fig. 1A, the striking fractionation of the  $\text{HCl}$  vs.  $\text{N}_2\text{O}$  correlation plot seen for October 1991 appears to represent a case intermediate between the SPADE (high aerosol) and STRAT (low aerosol) data. The initial volcanic material spreading out from the lower latitudes is known to have different chemical composition than the aerosol characteristic of the later decaying period, and it may be that the heterogeneous rates are even faster on the smaller, younger sulfate aerosol matter.

The observed dependence of  $\text{ClO}/\text{Cl}_y$  on aerosol surface concentration is also compared with model predictions in Fig. 4. These data have been selected to cover a narrow range of solar zenith angles (62-82°), and have been normalized to  $\text{O}_3$  amount with a small correction term generated from the observed dependence of  $\text{ClO}/\text{Cl}_y$  on measured ozone. The SPADE  $\text{ClO}$  data of May 1993 [Stimpfle *et al.*, 1994] are significantly lower than the 1991/2  $\text{ClO}$  data [Toohey *et al.*, 1993] for similar particle surface area concentrations of around  $2\text{-}6 \mu\text{m}^2/\text{cm}^3$ , due to higher  $\text{NO}_x$  values. These lower  $\text{ClO}$  data of SPADE were used [Stimpfle *et al.*, 1994] to infer (low)  $\text{ClONO}_2/\text{Cl}_y$  ratios, which were discrepant with

the direct  $\text{HCl}/\text{Cl}_y$  measurements of ALIAS. When dependence on  $\text{NO}_x$  is taken into account, the consistency between the model calculations and the measured  $\text{ClO}/\text{Cl}_y$  is very good. However, the plots of Fig. 4 illustrate that a discrepancy between  $\text{ClO}/\text{Cl}_y$  and  $\text{HCl}/\text{Cl}_y$  still remains, in that the temperatures needed to match the  $\text{HCl}/\text{Cl}_y$  observations are somewhat lower than those needed for  $\text{ClO}/\text{Cl}_y$ . Nevertheless, the model results do show that low  $\text{HCl}/\text{Cl}_y$  values of 40% can be expected in a high aerosol environment at only moderately low ( $\leq 207$  K) temperatures. Furthermore, these results are consistent with the results of a companion paper [Dessler *et al.*, 1997], which report HALOE satellite observations of a 26% increase in  $\text{HCl}/\text{Cl}_y$  from 1992 to 1996.

The observed dependence of lower stratospheric  $\text{HCl}$  concentration on surface area concentration over the five-year period following the volcanic eruption of Mt. Pinatubo demonstrates that aerosol chemistry at temperatures  $\geq 200$  K has a more pronounced effect on  $\text{Cl}_y$  partitioning than previously believed. Lower  $\text{HCl}/\text{Cl}_y$  ratios are associated with increased reactive chlorine, products of enhanced heterogeneous chemistry on sulfate aerosol which is responsible [Michelsen *et al.*, 1997] for the underestimation by atmospheric models [Solomon *et al.*, 1996] of observed ozone loss in volcanic aerosol conditions.

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## FIGURE CAPTIONS

Figure 1.  $\text{HCl}$  vs.  $\text{N}_2\text{O}$  from 1991 through 1996, compared with expected total inorganic chlorine ( $\text{Cl}_y$ ) mixing ratios calculated from the  $\text{N}_2\text{O}$  values [Woodbridge *et al.*, 1995]. The dashed lines are polynomial fits to the data, where the flights of Oct. 4 and 6 1991, and those of May 1 and 11 1993 are not included.

Figure 2. A. Superposition of the polynomial fits of Fig. 1 to the  $\text{HCl}/\text{Cl}_y$  vs.  $\text{N}_2\text{O}$  mixing ratios divided by  $\text{Cl}_y$ , showing the apparent time evolution. B. Measured  $\text{HCl}/\text{Cl}_y$  vs latitude over the range 40-60 °N for data constrained by: pressures  $\leq 67$  mbar,  $\text{N}_2\text{O} \leq 260$  ppbv,  $\text{ClO} \leq 150$  pptv, and  $1500 \leq \text{O}_3 \leq 2500$  ppbv. The solid lines are the average values over the given latitude range to illustrate the apparent evolution with time of  $\text{HCl}/\text{Cl}_y$  over this latitude range.

Figure 3. Measured  $\text{HCl}/\text{Cl}_y$  vs aerosol surface area concentration for several flights from December 8 1992 through July 22 1996. All data are constrained by: pressures  $\leq 67$  mbar,  $\text{N}_2\text{O} \leq 260$  ppbv, and  $\text{ClO} \leq 150$  pptv, and  $1500 < \text{O}_3 \leq 2500$  ppbv. The lower panel shows the flight-average values with one-sigma standard deviations from the mean, and an exponential least-squares fit to the flight-average points.

Figure 4. Comparison between theory and measurements. Flight-average values of  $\text{HCl}/\text{Cl}_y$  and  $\text{ClO}/\text{Cl}_y$  vs aerosol surface area are constrained by pressure  $\leq 67$  mbar,  $\text{N}_2\text{O} \leq 260$  ppbv,  $\text{ClO} \leq 150$  pptv, and  $1500 \leq \text{O}_3 \leq 2500$  ppbv. In addition to the flights of Fig. 3, flights of Oct. 4 and 6, 1991, and May 1 and 11, 1993 are included. Calculated values at 18 km are plotted for temperatures of 2.01, 204, and 230 K.

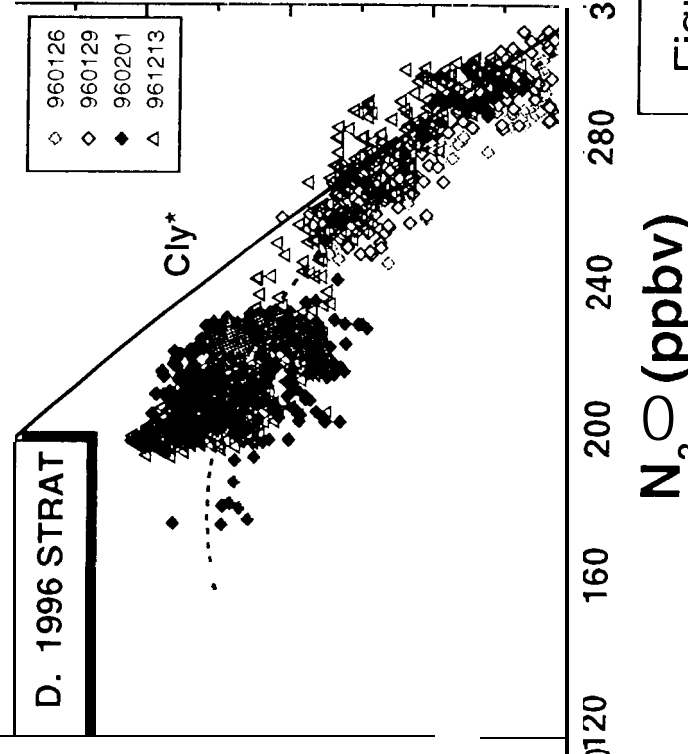
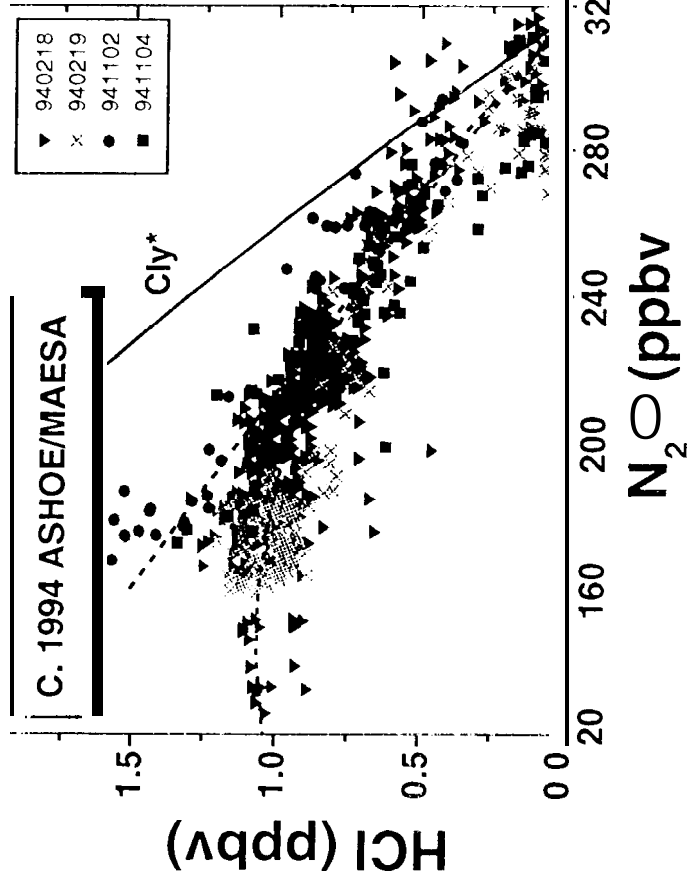
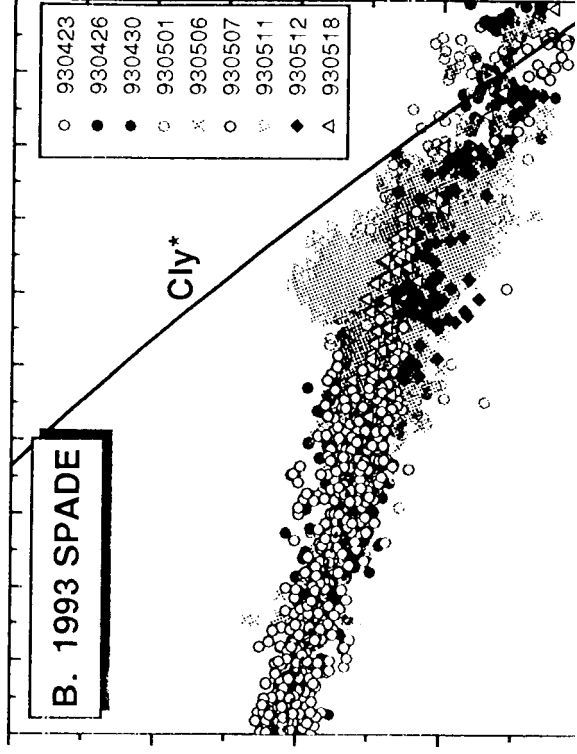
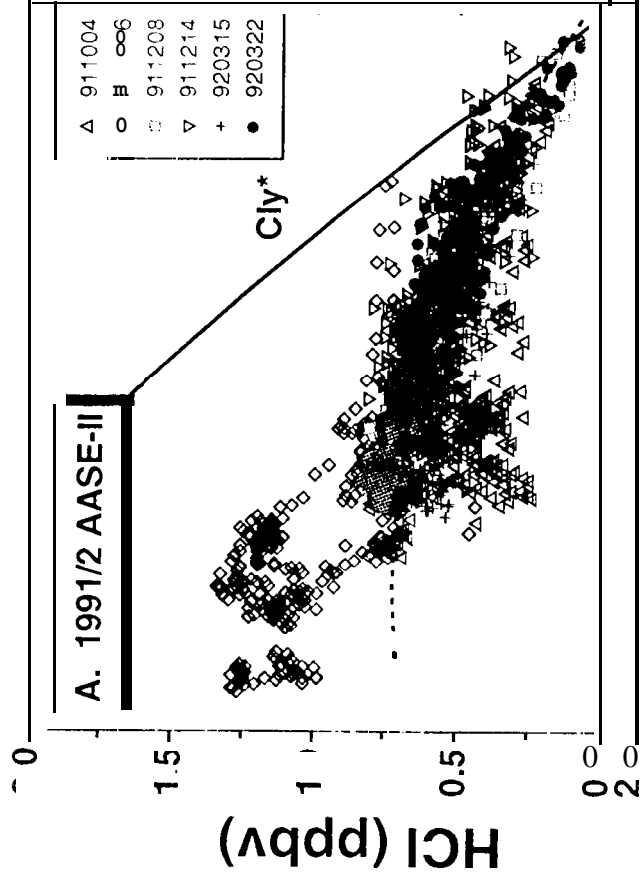


Figure 1

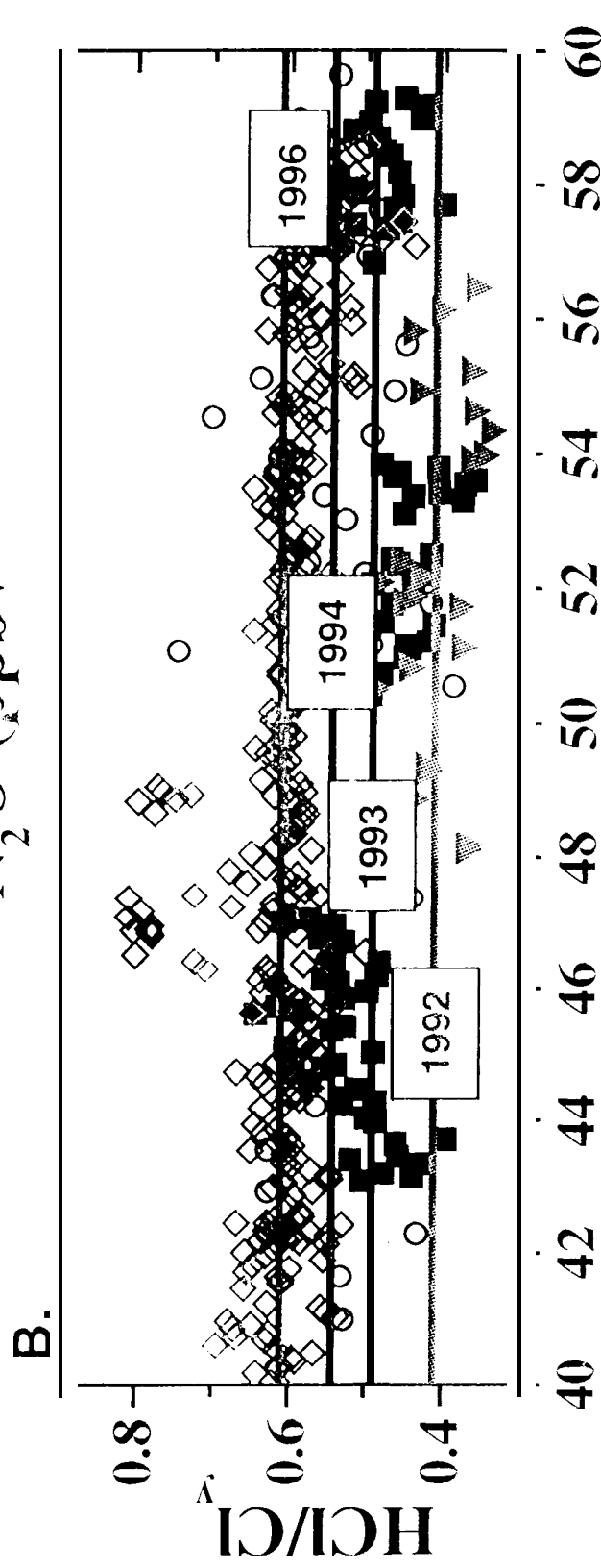
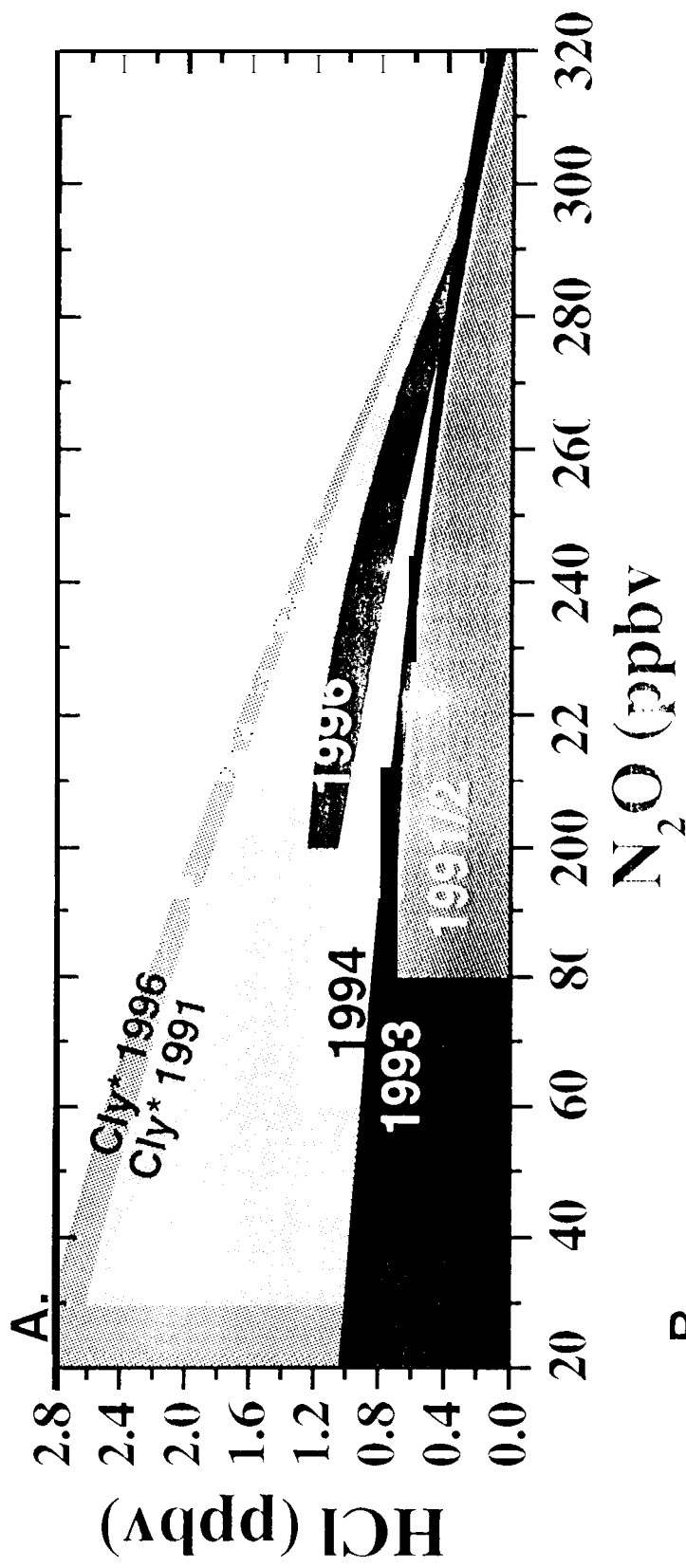
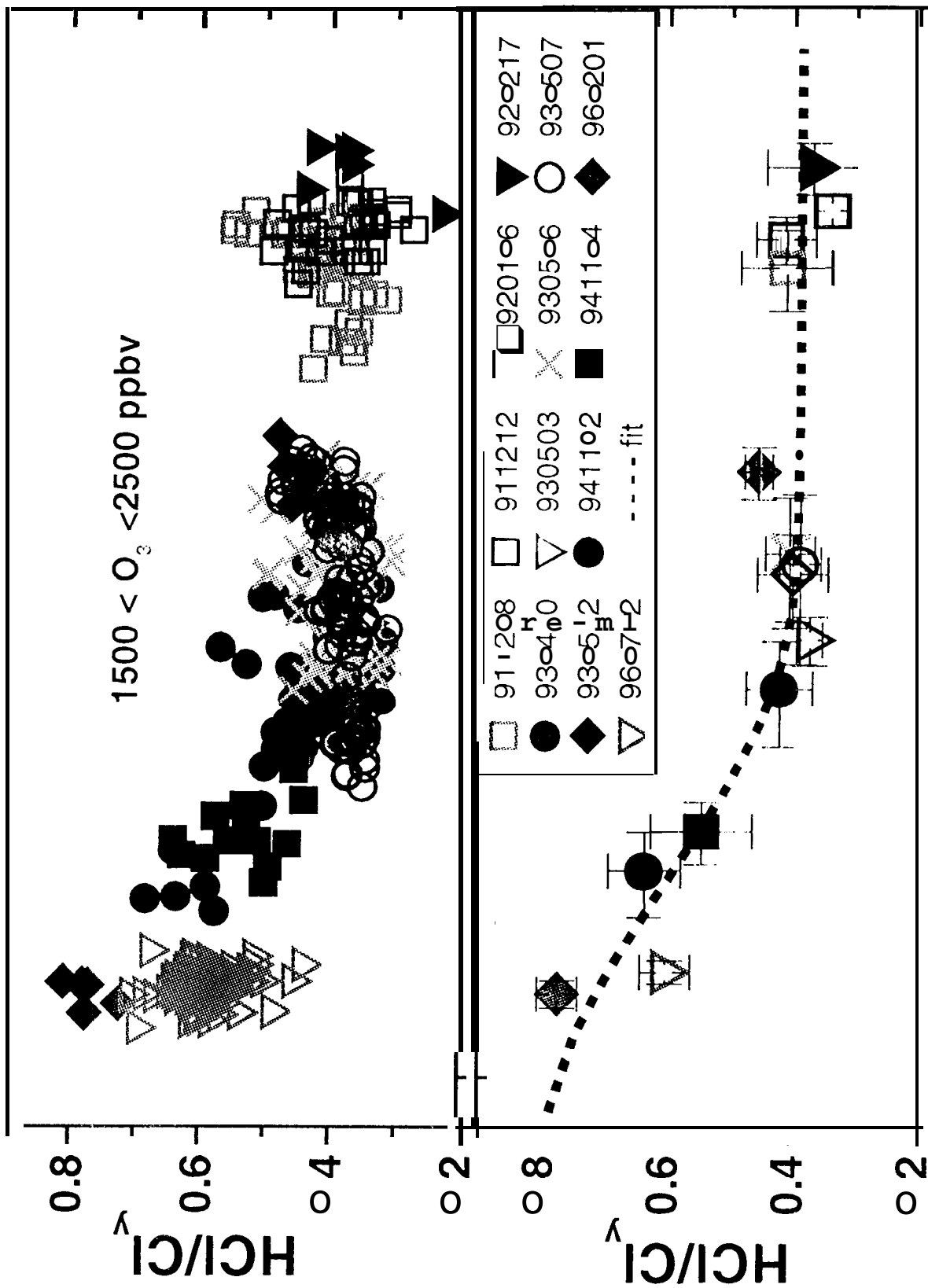


Figure 2



50

10<sup>2</sup> μm<sup>2</sup>/cm<sup>3</sup>

Figure 3

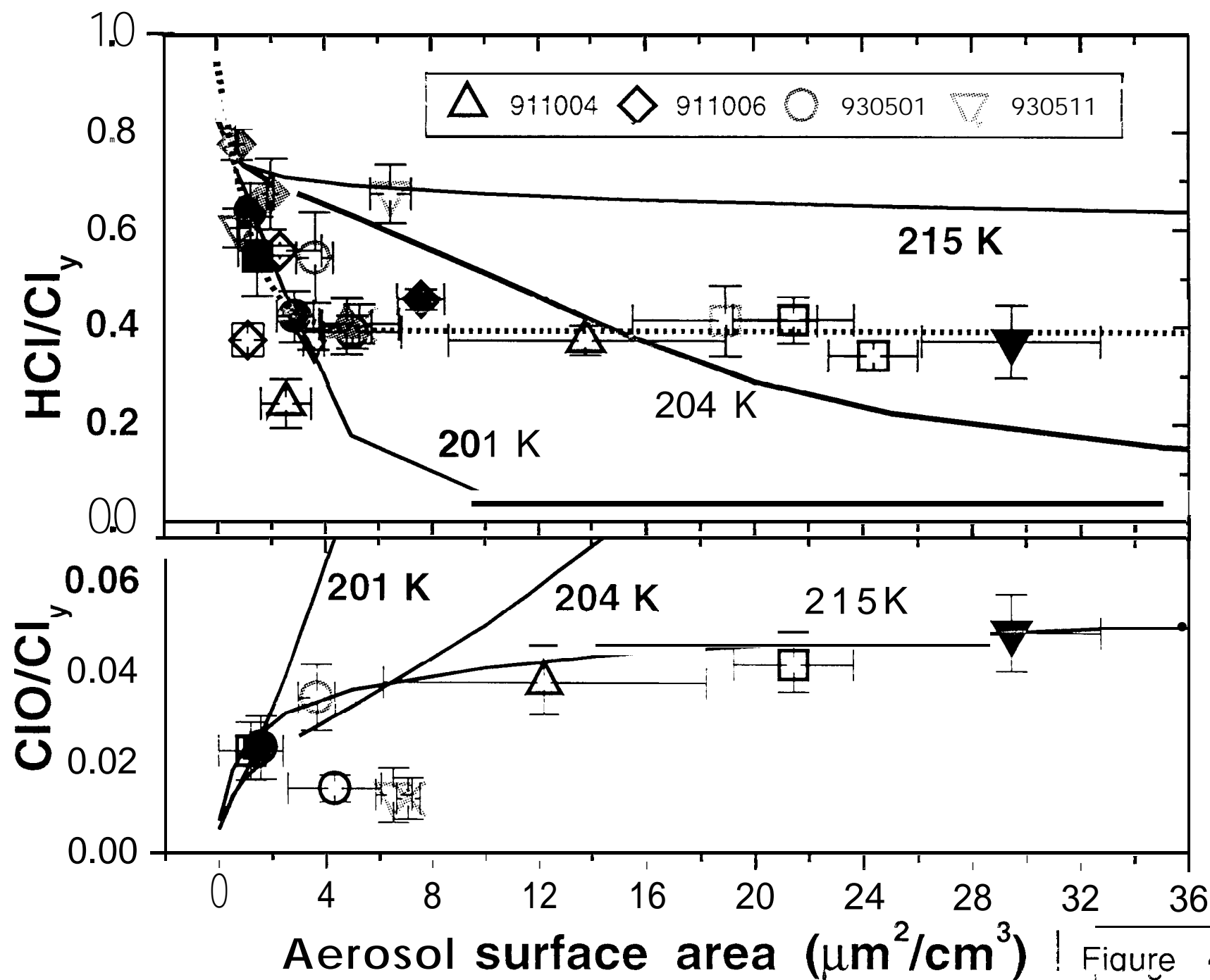


Figure 4